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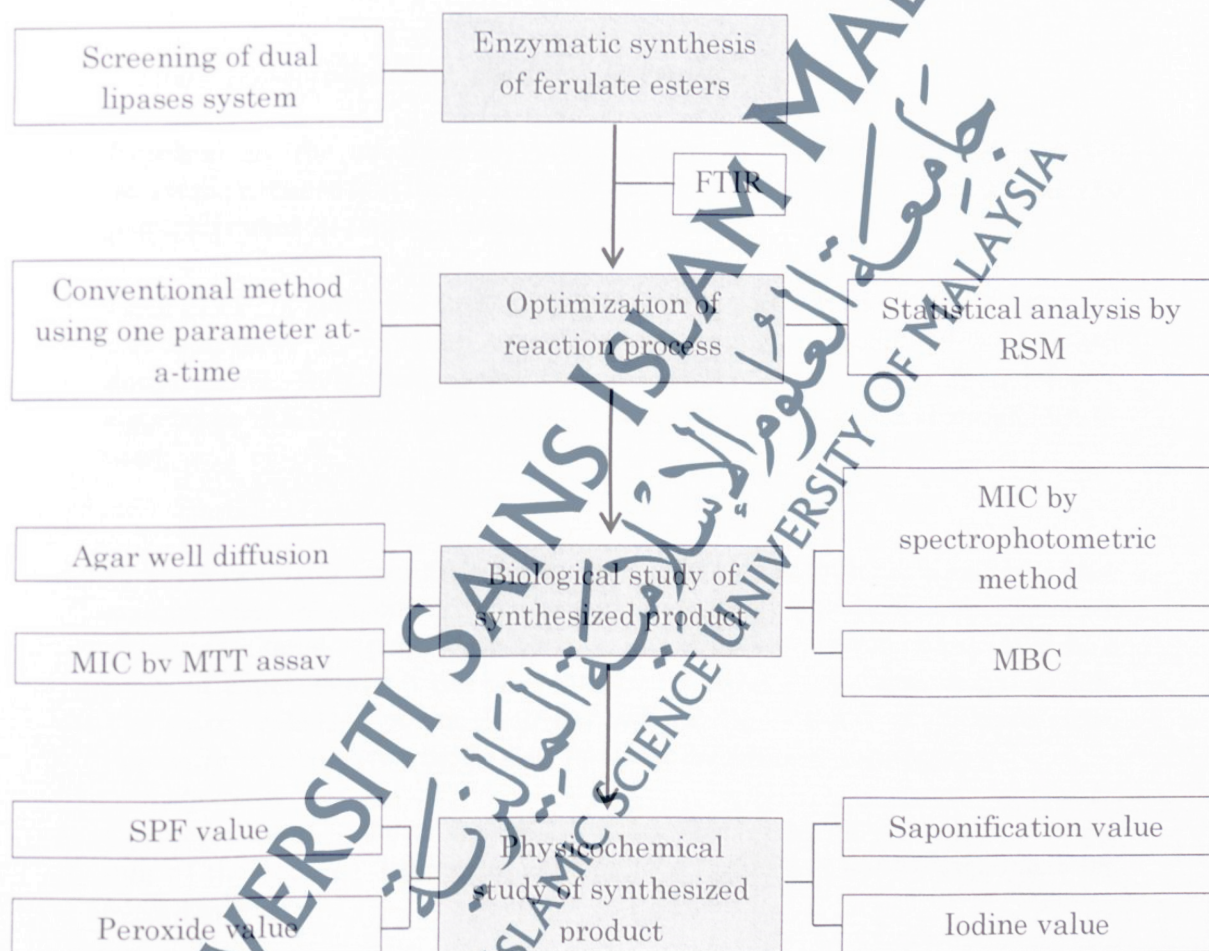
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APPENDICES

APPENDIX A

Experimental design



APPENDIX B

Formula for Statistical Analysis

(Design Expert Version 7.1.6, *User's Guide*)

1. **Model** represents the terms estimating factor effects.
2. **Residual** are the unexplained variation seen as the difference between the observed response and the value predicted by the model for a particular design point. It is used to estimate experimental error.
3. **Lack of fit** compares the residual error mean square (MS) to the pure error MS and represents the variation of data around the fitted model for a particular design point. It is used to estimate experimental error. If a model has a significant lack of fit it is not good predictor of the response and should not be used.
4. **Pure error** or experimental error is the normal variation in the response which appears when an experiment is repeated. Repeated experiments rarely produce exactly the same results. Pure error is the minimum variation expected in a series of experiments. It can be estimated by replicating points in the design. The more replicated points the better will be the estimate of the pure error. Pure error is used to test the lack of fit terms for possible significance.
5. **Corrected total** is the total sum of squares corrected for the mean. It is the sum of the squared differences between the individual observations and the overall average.
6. **Sum of squares (SS)** is the sum of the squared distances from the mean due to an effect.

6.1 **Model SS** is the sum of squares for terms in the model.

6.2 **Residual SS** is the sum of squares for all the terms not included in the model.

$$\text{Residual SS} = \text{Corrected total SS} - \text{Model SS}$$

6.3 **Lack of fit SS** is the residual SS after removing the pure error SS.

$$\text{Lack of fit SS} = \text{Corrected total SS} - \text{Pure error SS} - \text{Model SS}$$

6.4 **Pure error SS** is the pure error SS for replicated points.

6.5 **Corrected total** is the sum of squared deviations of each point from the mean.

7. **Degree of freedom (d.f)** is the number of independent comparisons available to estimate a parameter.

7.1 **Model d.f** is the model which comprises the number of model terms including the intercept minus one.

$$\text{Model d.f} = k + k(k + 1)/2$$

Where, K = Degree of freedom of first order coefficients

7.2 **Residual d.f** is the estimation of variance around the model.

$$\text{Residual d.f} = \text{Corrected total d.f} - \text{Model d.f}$$

7.3 **Lack of fit d. f** is the amount of information available from the replicated points.

$$\text{Lack of fit d.f} = \text{Residual d.f} - \text{Pure error d. f}$$

7.4 **Pure error d.f** is the amount of information available from replicated points.

$$\text{Pure error} = n_1 - 1$$

Where, n_1 = Total number of center points

7.5 **Corrected total d.f** is the total degrees of freedom for the experiment, minus one for the mean.

$$\text{Corrected total d.f} = n_1 + n_2 - 1$$

Where, n_1 = Total number of center points

n_2 = Total number of axial and fractional points

8. **Mean square (MS)** is the sum of squares divided by the number of degrees of freedom and it is used to estimate the variance.

8.1 **Model MS** is the estimate of model variance.

$$\text{Model MS} = \text{Model SS} / \text{Model d.f}$$

8.2 **Residual MS** is the estimate of process variance.

$$\text{Residual MS} = \text{Residual SS} / \text{Residual d.f}$$

8.3 **Lack of fit MS** is the estimate of lack of fit.

$$\text{Lack of fit MS} = \text{Lack of fit SS} / \text{Lack of fit d.f}$$

8.4 **Pure error MS** is the estimate of pure error variance.

$$\text{Pure error MS} = \text{Pure error SS} / \text{Pure error d.f}$$

9. **F-value** is a probability distribution used to compare variances by examining their ratio. If they are equal then the F value would equal 1. The F value in the ANOVA table is the ratio of model mean square (MS) to the appropriate error mean square. The larger the ratio, the larger the F value and the more likely that the variance contributed by the model is significantly larger than random error.

9.1 **F-value of model** compares model variance with residual variance.

$$\text{F-value of model} = \text{Model MS} / \text{Residual MS}$$

9.2 **F-value of lack of fit** compares lack of fit variance with pure error variance.

$$\text{F-value of lack of fit} = \text{Lack of fit MS} / \text{Pure error MS}$$

10. **Prob>F** is the probability value that is associated with the F Value for this term. It is the probability of getting an F Value of this size if the term did not have an effect on the response. In general, a term that has a probability value less than 0.05 would be considered a significant effect. A probability value greater than 0.10 is generally regarded as not significant.
11. **Coefficient of determination (R^2)** is an estimate of the overall variation in the data around the mean accounted for by the model. A value of 1.00 represents the ideal case at which 100 % of the variation in the observation value can be explained by the chosen model.

$$R^2 = 1 - [\text{Residual SS} / (\text{Model SS} + \text{Residual SS})]$$

APPENDIX C

IR Absorption Peaks for Several Functional Groups

Functional groups		Absorption peaks		
			Frequency (cm^{-1})	Wavelength (μm)
C-H	Alkane	Stretch	3000-2850	3.33-3.51
	-CH ₃	Bend	1450 and 1375	6.90 and 7.27
	-CH ₂	Bend	1465	6.83
	Alkene	Stretch	3100-3000	3.23-3.33
		Out of plane	1000-650	10.0-15.3
	Aromatic	Stretch	3150-3050	3.17-3.28
Out of plane		900-690	11.1-14.5	
C=C	Alkene		1680-1600	5.95-6.25
	Aromatic		1600 and 1475	6.25 and 6.78
C=O	Aldehyde		1740-1720	5.75-5.81
	Ketone		1725-1705	5.80-5.87
	Carboxylic acid		1725-1700	5.80-5.88
	Ester		1750-1730	5.71-5.78
	Amide		1670-1640	6.00-6.10
C-O	Alcohol, Ether, Ester, Carboxylic acid		1300-1000	7.69-10.0
O-H	Alcohol	Free	3650-3600	2.74-2.78
	H-bonded		3500-3200	2.86-3.13
	Carboxylic acid		3400-2400	2.94-4.17

APPENDIX D

McFarland Standard

BaCl ₂	H ₂ SO ₄	Cell/mL	Optical Density
0.1	9.9	3 x 10 ⁸	0.18-0.19
0.2	9.8	6 x 10 ⁸	0.30-0.32
0.3	9.7	9 x 10 ⁸	0.47-0.50
0.4	9.6	12 x 10 ⁸	0.60-0.61
0.5	9.5	15 x 10 ⁸	0.70-0.72
0.6	9.4	18 x 10 ⁸	0.88-0.90
0.7	9.3	21 x 10 ⁸	1.00-1.10
0.8	9.2	24 x 10 ⁸	1.20-1.25
0.9	9.1	27 x 10 ⁸	1.20-1.25
1.0	9.0	30 x 10 ⁸	1.30-1.35

APPENDIX E

Normalized Product Function in Calculation of SPF

Wavelength (λ nm)	EE x I (normalized)
290	0.0150
295	0.0817
300	0.2874
305	0.3278
310	0.1864
315	0.0839
320	0.0180
Total	1

APPENDIX F

List of Publications

- Salina Mat Radzi, **Nurul Jannah Abd Rahman**, Hanina Mohd Noor & Norlelawati Ariffin. 2014. "Enzymatic Synthesis of Olive-Based Ferulate Esters: Optimization by Response Surface Methodology". *International Journal of Chemical, Nuclear, Metallurgical and Materials Engineering*. Vol. 8. (8): p. 654-657.
- Nurul Jannah Abd Rahman**, Salina Mat Radzi & Hanina Mohd Noor. 2015. "Dual Lipases System in Transesterification of Ethyl Ferulate with Olive Oil: Optimization by Response Surface Methodology". *Asian Journal of Applied Sciences*. Vol. 3 (1).
- Nurul Jannah Abd Rahman**, Salina Mat Radzi, Hanina Mohd Noor & Norlelawati Ariffin. 2015. "Antibacterial Study of Olive-Based Ferulate Esters". *World Journal of Pharmacy and Pharmaceutical Sciences*. Vol. 4 (1).
- Salina Mat Radzi, **Nurul Jannah Abd Rahman**, Hanina Mohd Noor & Mahiran Basri. 2014. "High Yield Synthesis of Kojic Ester Using Dual Enzymes System and Their Antibacterial Activity". *Key Engineering Materials*. Vol. 594-595. p. 362-369.
- Nurul Jannah Abd Rahman** & Salina Mat Radzi. 2013. "Optimization via Response Surface Methodology (RSM) of Dual Enzymes System for the Production of Ferulate Esters". *Abstract of 38th Annual Conference of the Malaysian Society for Biochemistry and Molecular Biology*.
- Nurul Jannah Abd Rahman**, Salina Mat Radzi & Hanina Mohd Noor. 2014. "The Use of Dual Lipases System in the Enzymatic Synthesis of Ferulate Esters: Modeling and Optimization by RSM". *Extended Abstract of The Regional Fundamental Science Congress*.
- Nurul Jannah Abd Rahman**, Salina Mat Radzi & Hanina Mohd Noor. 2014. "Antimicrobial Study of Ferulate Esters from Transesterification Reaction between Ethyl Ferulate and Olive Oil". *Extended Abstract of The 6th International Conference on Postgraduate Education*.