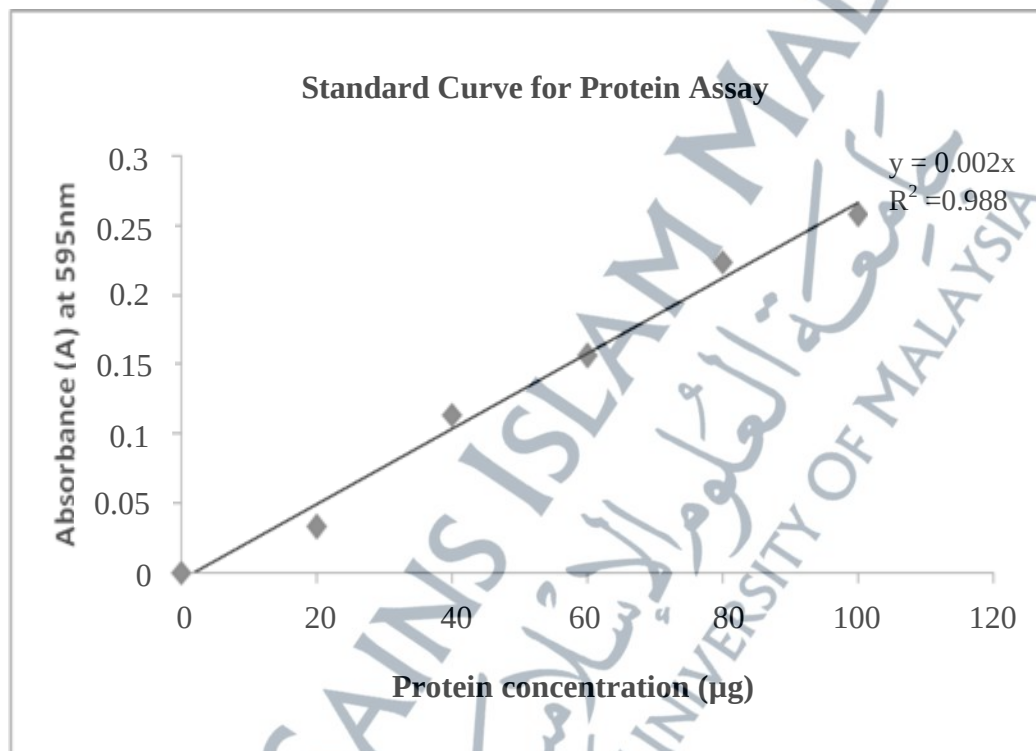


APPENDIX A

THE STANDARD CURVE OF ABSORBANCE VERSUS CONCENTRATION OF BOVINE SERUM ALBUMIN (BSA)



APPENDIX B

PREPARATION OF BRADFORD REAGENT, BOVINE SERUM ALBUMIN (BSA) STANDARD SOLUTION AND DISTILLED WATER IN PROTEIN

ASSAY PROCEDURE

Tube	BSA ^a (μL)	Distilled water (μL)	Bradford Reagent (mL)
1	0	100.0	5.0
2	20.0	80.0	5.0
3	40.0	60.0	5.0
4	60.0	40.0	5.0
5	80.0	20.0	5.0
6	100.0	0	5.0
Sample ^b (μL)			
Free lipase	50.0	50.0	5.0
Immobilized lipase (before and after immobilization)	50.0	50.0	5.0

Sample^a = 1mg/mL of BSA concentration was prepared and used as a standard protein

Sample^b = 50 μL was used in the determination of protein content

APPENDIX C

CALCULATION OF SPECIFIC ACTIVITY

$$\text{Specific Activity (mmol.min/mg)} = \frac{[(V_{\text{control}} - V_{\text{sample}}) M] / t}{W}$$

Whereby,

V_{control} = titration volume of NaOH for control (mL)

V_{sample} = titration volume of NaOH for sample (mL)

M = molarity of NaOH

t = reaction time

W = amount of protein (mg)

APPENDIX D

CALCULATION OF RELATIVE ACTIVITY

The relative activity was determined as percentage of yield of certain activities compared to the activity of reaction at maximum yield by using the following equation:

$$\text{Relative activity (\%)} = \frac{\text{Specific activity in treatment}}{\text{Maximum yield specific activity}} \times 100$$

APPENDIX E

CALCULATION THE PERCENTAGE CONVERSION OF ESTER

The percentage conversion of ester was determined by using the following equation:

$$\% \text{ conversion} = [(V \text{ control} - V \text{ sample}) / V \text{ control}] \times 100$$

Whereby,

V_{control} = titration volume of NaOH for control (mL)

V_{sample} = titration volume of NaOH for sample (mL)

APPENDIX F

CALCULATION THE PERCENTAGE OF IMMOBILIZATION AND PROTEIN LOADING

The immobilized enzyme was calculated in term of percentage of immobilization (%) by using the equation as stated below:

$$\% \text{ immobilization} = \frac{X_b - X_a}{X_b} \times 100$$

Whereby,

X_a = Amount of lipase after immobilization (*from standard curve*)

X_b = Amount of lipase before immobilization (*from standard curve*)

$$\text{Protein loading } (\mu\text{g/g}) = \frac{X_b - X_a}{A}$$

A = initial mass of support used

APPENDIX G

CALCULATION OF THE REACTION RATE, V (mmol/L/min/mg)

$$V = \frac{V_{\max} [\text{But}][\text{Ol}]}{K_{\text{m}(\text{But})}[\text{Ol}](1 + [\text{Ol}]/K_{\text{i}(\text{Ol})}) + K_{\text{m}(\text{Ol})}[\text{But}] + [\text{But}][\text{Ol}]}$$

$$V = \frac{V_{\max} [\text{But}][\text{Ol}]}{K_{\text{m}(\text{Ol})}[\text{But}](1 + [\text{But}]/K_{\text{i}(\text{But})}) + K_{\text{m}(\text{Ol})}[\text{Ol}] + [\text{But}][\text{Ol}]}$$

$$V = \frac{V_{\max} [\text{But}][\text{Ol}]}{K_{\text{m}(\text{But})}[\text{Ol}](1 + [\text{Ol}]/K_{\text{i}(\text{Ol})}) + K_{\text{m}(\text{Ol})}[\text{But}](1 + [\text{But}]/K_{\text{i}(\text{But})}) + [\text{But}][\text{Ol}]}$$

Whereby,

V = initial reaction rate

[But] = concentration of butanol

[Ol] = concentration of oleic acid

V_{max} = maximum velocity

K_{m(But)}, K_{m(Ol)} = Michaelis constant of butanol, oleic acid

K_{i(But)}, K_{i(Ol)} = inhibition constant of butanol, oleic acid